

## The Stereochemistry of Some Stages in Gibberellin Biosynthesis

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**Summary** The stereospecific labelling of the gibberellins and the kaurenolides with 2(*R*) and 5(*R*)-[<sup>3</sup>H]mevalonate leads to the conclusion that the dehydrogenation of ring A is a *cis*-process and that hydroxylation of ring B takes place with retention of configuration.

It has been shown<sup>1</sup> that the biosynthesis of the gibberellins and kaurenolides lies through the tetracyclic diterpene (–)-kaurene (I). This biosynthesis is followed by

stepwise hydroxylation leading to the formation of 7β-hydroxy-(–)-kaur-16-en-19-oic acid which acts as a precursor of both the kaurenolides and the gibbane aldehyde (II).<sup>2</sup> The absolute stereochemistry of the mevalonodi-hydrogen atoms of the open-chain terpenoid precursors has been defined<sup>3</sup> and we now extend this information to gibberellin and kaurenolide biosynthesis.

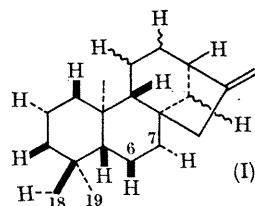
The formation of the antipodal (–)-kaurene is a function of the relative orientation of the double bonds of a geranylgeraniol precursor but does not disturb the pro “*R*” or pro

### *The incorporation of mevalonic acid into the metabolites of Gibberella fujikuroi*

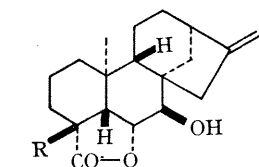
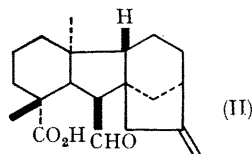
	2( <i>R</i> )- <sup>3</sup> H		5( <i>R</i> )- <sup>3</sup> H	
	<sup>3</sup> H : <sup>14</sup> C	No. of <sup>3</sup> H	<sup>3</sup> H : <sup>14</sup> C	No. of <sup>3</sup> H
Mevalonic acid .. .. .	9·0 : 1	–	9·5 : 1	–
(–)-Kaurene (I) .. .. .	9·0 : 1	4	9·5 : 1	4
7-Hydroxykaurenolide (III) .. .. .	9·0 : 1	4	9·4 : 1	4
7,18-Dihydroxykaurenolide (IV) .. .. .	8·0 : 1	3·6	9·15 : 1	4
Gibberellic acid (V)† .. .. .	6·6 : 1	3	4·96 : 1	2
Gibberellin A <sub>4</sub> (VI)† .. .. .	7·0 : 1	3	–	–
Gibberellin A <sub>13</sub> (VII)† .. .. .	6·8 : 1	3	7·43 : 1	3

† Purified as their methyl esters.

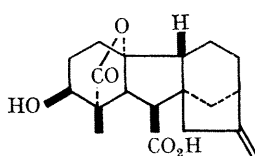
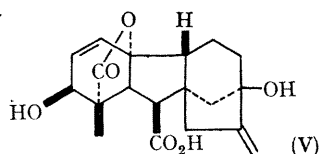
"S" character of the individual hydrogen atoms. On the basis of analogy with the steroids, those that would be expected to originate from [2(R), 4(R), and 5(R)-<sup>3</sup>H]mevalonic acid are shown in (I). We have already described<sup>4</sup>



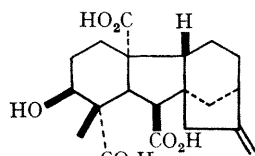
H = label from 2(R)-, 4(R)-, or 5(R)-mevalonate



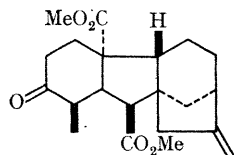
(III) R=Me (IV) R=CH<sub>2</sub>OH



(VI)



(VII)



(VIII)

our results with doubly-labelled [4(R)-<sup>3</sup>H;2-<sup>14</sup>C]mevalonic acid and we now detail our results with [2(R)-<sup>3</sup>H;2-<sup>14</sup>C]- and [5(R)-<sup>3</sup>H;2-<sup>14</sup>C]mevalonic acid, the latter fed as its DBED salt.

7-Hydroxykaurenolide from the (2(R)-<sup>3</sup>H] experiment

was oxidized to 7-oxokaurenolide [<sup>3</sup>H:<sup>14</sup>C, 7·2:1] whilst the 7,18-dihydroxykaurenolide was oxidized to 7-oxo-18-norkaurenolide [<sup>3</sup>H:<sup>14</sup>C, 5·0:0·75] thus locating a mevalonoid label at positions 7 $\alpha$  and 18. Methyl gibberellate was converted into methyl gibberate [<sup>3</sup>H:<sup>14</sup>C, 6·68:1] with no loss of label.

The metabolites from the [5(R)-<sup>3</sup>H] experiment were degraded as follows. 7-Hydroxykaurenolide was oxidized to 7-oxokaurenolide [<sup>3</sup>H:<sup>14</sup>C, 9·77:1] and the latter treated with base. The gummy product showed [<sup>3</sup>H:<sup>14</sup>C, 7·25:1]. Alternatively, the kaurenolide was converted<sup>5</sup> into 6-oxo(-)-kaur-16-en-19-oic acid [<sup>3</sup>H:<sup>14</sup>C, 7·35:1] thus locating a label at the 6 $\alpha$ -position. Gibberellin A<sub>13</sub> trimethyl ester was oxidized to the corresponding ketone [<sup>3</sup>H:<sup>14</sup>C, 7·43:1] and the latter treated with alkali and then remethylated with diazomethane to form the nor-gibbane (VIII) [<sup>3</sup>H:<sup>14</sup>C, 4·85:1].<sup>6</sup> Methyl gibberellate was converted into methyl allogibberate [<sup>3</sup>H:<sup>14</sup>C, 4·99:1] and thence to methyl gibberate [<sup>3</sup>H:<sup>14</sup>C, 4·96:1], which was oxidized to methyl gibberdionate [<sup>3</sup>H:<sup>14</sup>C, 2·66:1],<sup>7</sup> thus locating a label at (the gibbane) position 11 in gibberellic acid. Hydrolysis of methyl gibberate with alkali gave an acid [<sup>3</sup>H:<sup>14</sup>C, 2·48:1] with the loss of only one tritium from the enolizable position of the cyclopentanone. Thus there was no evidence for the presence of a label at the 10-position.

A number of conclusions may be drawn from these results. Hydroxylation of ring B to form the kaurenolides must take place with retention of configuration at C-6 and C-7. However, ring-contraction to form the gibberellins which takes place at the aldehyde oxidation level, results in the loss of 5(R)-mevalonoid hydrogen from C-6 (C-10 of the gibbane skeleton). This suggests that the leaving group which initiates ring contraction possesses the 6 $\beta$ -stereochemistry. Secondly there is evidence for the stereochemistry of processes on ring A of gibberellic acid. The dehydrogenation step to form the  $\Delta^3$ -double bond involves a "cis" elimination of hydrogen from the " $\alpha$ " face of a saturated gibberellin. This must exclude processes involving hydroperoxidation of  $\Delta^2$ -gibberellins as this would require the elimination of a  $\beta$ -hydrogen atom (cis to the hydroperoxide) at C-4.

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<sup>1</sup> B. E. Cross, R. H. B. Galt, and J. R. Hanson, *J. Chem. Soc.*, 1964, 295.

<sup>2</sup> J. R. Hanson and A. F. White, *Chem. Comm.*, 1969, 410.

<sup>3</sup> G. Popják and J. W. Cornforth, *Biochem. J.*, 1966, **101**, 553.

<sup>4</sup> J. R. Hanson and A. F. White, *J. Chem. Soc. (C)*, 1969, 981.

<sup>5</sup> B. E. Cross, R. H. B. Galt, and J. R. Hanson, *J. Chem. Soc.*, 1963, 2944.

<sup>6</sup> R. H. B. Galt, *J. Chem. Soc.*, 1965, 3143.

<sup>7</sup> B. E. Cross, *J. Chem. Soc.*, 1954, 4670.